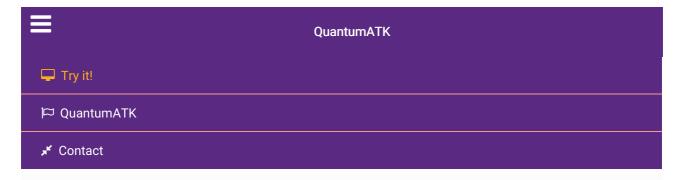
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How to Setup Basic Molecular Dynamics Simulations

How to Setup Basic Molecular Dynamics Simulations

Version: U-2022.12

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Molecular Dynamics

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md_tutorial_npt_300k_1bar.hdf5

In this tutorial, you will learn how to setup basic molecular dynamics (MD) simulations and analyze the resulting trajectories using the NanoLab GUI in **QuantumATK** . At the end of this tutorial, you will be familiar with simulating systems in a microcanonical ensemble (NVE), canonical ensemble (NVT) and isothermal-isobaric ensemble (NPT), using **QuantumATK**.

Pre-requisites

- Basic familiarity with NanoLab and atkpython, such as from the Introduction to the Workflow Builder and How to use QuantumATK for fast and easy atomic scale modelling
- Basic familiarity with Molecular Dynamics simulations. If needed, you can read our technical note on Molecular Dynamics for a basic introduction to MD along with best practices, tips and tricks when setting up and analyzing MD simulations.



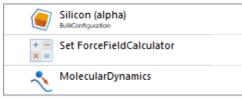
NVE Simulations

Let us setup a microcanonical ensemble simulation in which the number of particles (N), volume of the simulation cell (V) and energy of the system (E) remains constant. Import the geometry we will need, silicon_bulk.py, to your Builder stash.

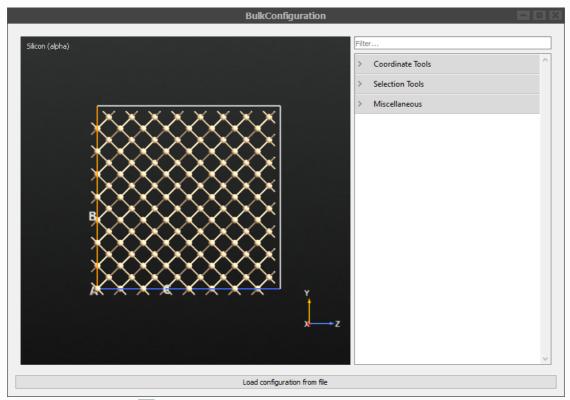
Workflow

- 1. Select the silicon supercell in Builder stash.
- 2. Click on the __icon and send the bulk configuration to the Workflow Builder
- 3. In the **Workflow Builder**, change the output **Filename** at the bottom of the middle panel to md-nve-1fs.hdf5.
- 4. Add the following blocks to the **Build** panel by double clicking on the corresponding icons in the **QuantumATK** tab in the right-hand panel:
 - Under Calculators: Set ForceFieldCalculator
 - Under Optimization and Dynamics: 🔨 Molecular Dynamics

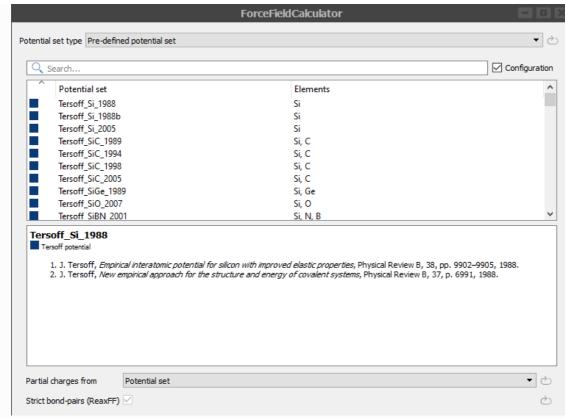
The resulting workflow will look like the below picture



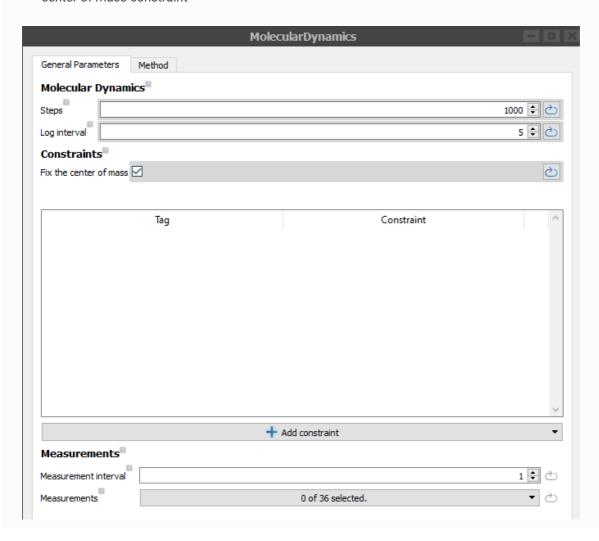
5. Double click the *Silicon (alpha)* block to visualize the bulk supercell. This is the system we will run MD for.



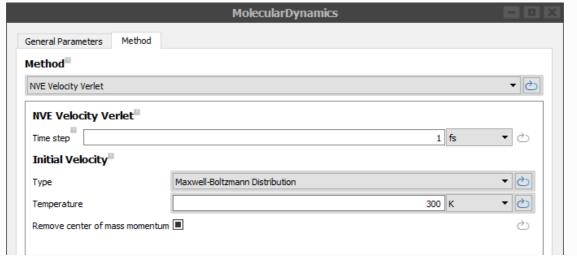
6. Double-click on the Set ForceFieldCalculator block and select the Tersoff_Si_1988 parameter set as shown below



7. Double-click on the *MolecularDynamics* block, set the **Steps** to *1000*, **Log interval** to *5* and set *Fix the center of mass* constraint



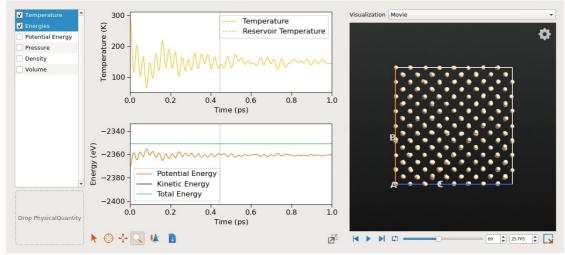
8. Go to **Methods** tab and choose **Method** as *NVE Velocity Verlet* from the pull down menu, set **Initial velocity Type** to *Maxwell-Boltzmann Distribution* and **Temperature** to *300 K*.



9. Send the workflow to the **Job Manager** as a script using and save the script as md-nve-1fs.py. This script runs in a few seconds, it can run locally.

Analyzing the Results

Once the simulation is finished, select the <code>md-nve-1fs.hdf5</code> file from the project folder and double-click the <code>MDTrajectory_0</code> object from the Data View. The trajectory will open in the <code>Movie Tool</code> plugin. Using the <code>Movie Tool</code> plugin, you can investigate the temperature as well as the kinetic, potential, and total energies of the system as a function of simulation time. Simultaneously, an animation of the motion of the atoms is displayed.



If you take a look at the temperature, you will observe the temperature dropping immediately and then oscillate around the set temperature. The reason is that the displayed temperature is the instantaneous temperature of the system, calculated from the kinetic energy of the atoms, via $E_{kin}(t)=3/2Nk_BT(t)$. Since the atoms are not free but interact with the neighboring atoms in the lattice, their initial kinetic energy will be partially converted into potential energy. For an *NVE* ensemble, a crucial parameter to monitor is the conservation of the total energy and find that the total energy curve in the **Movie Tool** remains constant with only small oscillations which are acceptable.

Tip

Try changing some MD parameters and see how they affect dynamics. For more tips, refer to

NVT Simulations

In NVT, we keep the temperature constant instead of the energy as in the NVE case. Download the workflow $\frac{1}{2}$ md_tutorial_nvt_300k.hdf5 to your project folder and open it in the Workflow Builder Tool in the Nanolab.

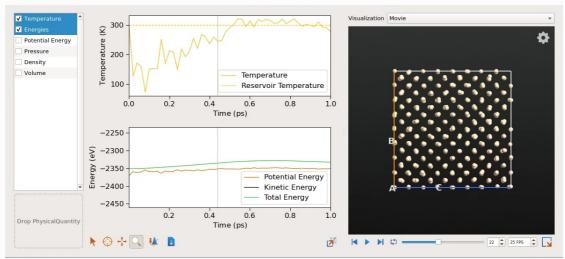
Workflow

The workflow should look similar to the NVE case, but the Molecular Dynamics block should list *NVT* as the **MD** method and a reservoir temperature of *300 K* will be listed. In this example, we use the *Nose-Hoover* type thermostat coupling to the reservoir. For more details on this method and other available thermostats, refer to the technical note on Molecular Dynamics.

After loading the workflow, export it as a script and submit the job.

Analyzing the Results

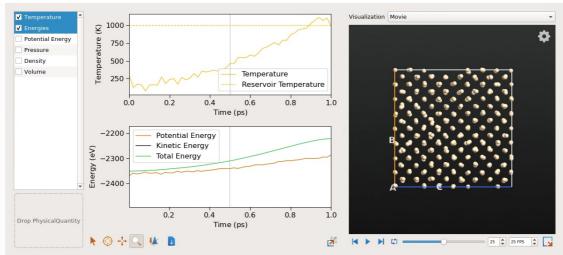
After the simulation is completed, open the NVT MD trajectory object with the **Movie Tool** and take a look at the temperature curve.



You will find that after a short initial equilibration period the curve fluctuates slightly around the selected reservoir temperature of 300 K. In fact, it does not look very different from the results of the NVE simulation in the previous section. In contrast to the NVE simulation, however, the final equilibrium temperature of the system is independent of the initial value, set via the Maxwell-Boltzmann distribution, and is instead coupled to the reservoir temperature. Let us take a closer look at this now.

Download the workflow $\frac{1000 \, \text{K}}{1000 \, \text{K}} = \frac{1000 \, \text{K}$

On opening the resulting trajectory in the movie tool, we see the temperature curve starting at an initial value of 300 K (as selected via the *Maxwell-Boltzmann distribution*).



Subsequently, as the system is in contact with the heat bath, the temperature quickly increases towards the reservoir temperature of 1000 K. You may also notice an increase in the total energy. In contrast to *NVE* simulations, this is not a sign of poorly chosen parameters, but reflects the fact that energy is exchanged with the virtual heat bath.

During the simulation at 1000 K, the motion of the atoms increases compared to the previous simulation at 300 K.

300 K 1000 K





If you increase the reservoir temperature to even higher values (e.g 3000 K), you will find that the crystal will begin to melt and become partially amorphous. Annealing crystals at high temperatures is in fact a common way to generate amorphous structures of a material. You can learn more about this method in the tutorial Generating Amorphous Structures.

Warning

Setting too high a temperature on a minimum energy configuration might result in the generation of very large forces on the atoms which could destabilize the system. Instead, slowly heat up the system from room temperature using a suitable *heating rate*.

NPT Simulations

In this ensemble, we keep the number of particles, pressure and temperature constant. Download the workflow $\frac{1}{2}$ md_tutorial_npt_300k_1bar.hdf5 to your project folder and open it in the Workflow Builder tool in the NanoLab.

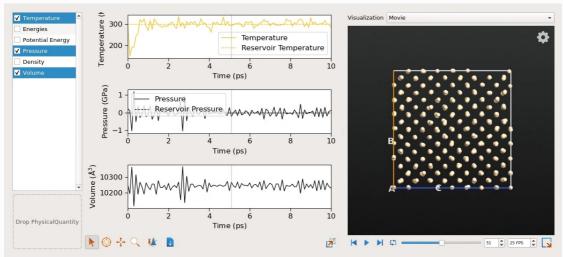
Workflow

The main difference you will observe in the MD block is the choice of NPT method. We use **NPT Martyna Tobias Klein** barostat in this example. For more details about this method and other NPT methods, refer to Molecular Dynamics. As you will notice the **reservoir temperature** and **pressure** are set to *300 K* and *1 bar*, respectively and an *isotropic coupling* is selected. This is because are simulating the system at standard temperature and pressure conditions and also allow the system to relax equally on all three lattice directions. Anisotropic coupling can be enabled to choose to fix some simulation cell boundaries.

Export the workflow as script and submit the job.

Analyzing the Results

Opening the trajectory with the **Movie Tool**, you can monitor the changes in the size and shape of the cell during the simulation. The temperature fluctuates around the selected reservoir temperature of 300 K.



To inspect the volume fluctuations you can select the *volume* measurement in the **Movie Tool**. The cell volume will show pronounced oscillations. These oscillations are a typical outcome of the *Nose-Hoover-like* temperature and pressure control. As long as their magnitude stays within reasonable boundaries, and does not increase dramatically throughout the simulation, there is no need to worry. The period of these oscillations can be changed by the *Barostat timescale* setting. The amplitude and frequency of the oscillations increases with decreasing barostat timescale.

The pressure during the simulation, seen by selecting the *Pressure* measurement in the **Movie Tool**, shows a similar behavior. It oscillates around the target pressure of 1 bar, but the amplitude is large. Such large fluctuations of the instantaneous pressure values are in fact not unusual in MD simulations, as the pressure is calculated based on the stress, which varies considerably with small changes of the volume. Moreover, pressure is a macroscopic property, and for such a small system it is more reasonable to consider the corresponding average rather than the instantaneous values.

Conclusion

You have learnt how to use the Workflow Builder tool to set up basic molecular dynamics simulations of NVE, NVT and NPT ensembles. You also know how to view the resulting trajectories using the Movie tool and interpret the results.

Feel free to change the MD settings and run the simulations to see how the dynamics changes and to familiarize yourself with the method.

You can try adding multiple MD blocks within a workflow to simulate the system in different ensembles in a sequential manner. For more inspiration, checkout predefined MD templates in the Templates section of the Workflow Builder.



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